

Prblm - A vapour absorption refrigeration system works with generator, ambient & evaporator temps. as 360 K, 310 K & 260 K resp. Find the COP of system. If evaporator temp. falls to 250 K, what should be the generator temp. in order to operate the system with same COP.

Soln -

$$\text{COP} = \left(\frac{360 - 310}{360} \right) \left(\frac{260}{310 - 260} \right)$$

$$= 0.138 \times 5.2$$

$$= 0.7176 \text{ Ans.}$$

$$\text{COP}' = \frac{360 - 310}{360}$$

$$\text{COP} = \left(\frac{T_G - 310}{T_G} \right) \left(\frac{250}{310 - 250} \right)$$

$$0.72 = 4.16 \left(\frac{T_G - 310}{T_G} \right)$$

$$0.72 T_G = 4.16 T_G - 1289.6$$

$$T_G = 375 \text{ K Ans.}$$

Prblm - A vapour absorption refrigeration system works with three thermal reservoirs. A RE of 100 W is required at 250 K, heat source is available at 400 K, heat rejection occurs at 300 K,

Find the minimum value of heat required in watt(W).

Solⁿ - $RE = 100 \text{ W}$

$$T_R = 250 \text{ K}$$

$$T_G = 400 \text{ K}$$

$$T_0 = 300 \text{ K}$$

$$COP = \frac{RE}{Q_G}$$

for min Q_G , COP should be max.

$$COP_{max} = \left(\frac{T_G - T_0}{T_G} \right) \left(\frac{T_R}{T_0 - T_R} \right)$$

$$\frac{RE}{Q_G} = \left(\frac{T_G - T_0}{T_G} \right) \left(\frac{T_R}{T_0 - T_R} \right)$$

$$\frac{100}{Q_G} = \left(\frac{400 - 300}{400} \right) \left(\frac{250}{300 - 250} \right)$$

$$Q_G = 80 \text{ W Ans.}$$