

# ADENOSINE TRIPHOSPHATE: THE ENERGY CURRENCY

- The energy in food does not transfer directly to cells for biologic work. Rather, energy from macronutrient oxidation is harvested and funneled through the energy-rich compound adenosine triphosphate (ATP).
- In essence, the energy donor–energy receiver role of ATP represents the cells’ two major energy transforming activities:
  1. Extract potential energy from food and conserve it within the bonds of ATP
  2. Extract and transfer the chemical energy in ATP to power biologic work
- ATP serves as the ideal energy-transfer agent. It “traps” within its phosphate bonds a large portion of the original food molecule’s potential energy. ATP also readily transfers this trapped energy to other compounds to raise them to a higher activation level.
- The cell contains other high-energy compounds (e.g., phosphoenolpyruvate; 1,3, diphosphoglycerate; phosphocreatine), but ATP is the most important.
- The bonds that link the two outermost phosphates (symbolized) represent high-energy bonds because they release considerable useful energy during hydrolysis. A new compound, adenosine diphosphate (ADP) forms when ATP joins with water, catalyzed by the enzyme adenosine triphosphatase (ATPase).



- This reaction generates considerable free energy, making ATP known as a high-energy phosphate compound. Infrequently, additional energy releases when another phosphate splits from ADP.
- The energy liberated during ATP breakdown directly transfers to other energy-requiring molecules.
- In muscle, the energy stimulates specific sites on the contractile elements to activate the molecular motors that power muscle fibers to shorten. Energy from ATP hydrolysis powers all forms of biologic work; thus, ATP constitutes the cell's "energy currency."

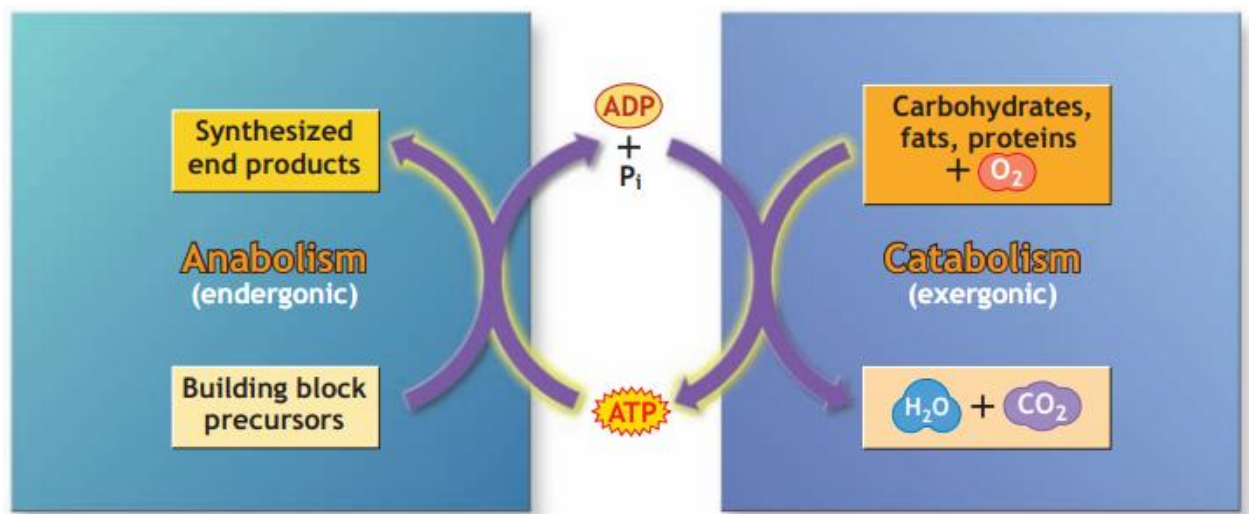


Figure 6.3 Catabolism–anabolism interactions. Continual recycling of ATP for biologic work from intracellular ADP, Pi, and energy released from stored macronutrients.

- ATP splits almost instantly without oxygen. This capability to hydrolyze ATP anaerobically to generate rapid energy transfer would not occur if energy metabolism required oxygen at all times.

- Bodily movements requiring this type of energy include sprinting 10 seconds for a bus, lifting an object, swinging a golf club, spiking a volleyball, or performing a pull-up or push-up.
- The body maintains a continuous ATP supply through different metabolic pathways:  
Three reactive processes that harness cellular energy to generate ATP aerobically—
  - the citric acid cycle,
  - B- oxidation,
  - respiratory chain—reside within the mitochondria.

## ATP: A Limited Currency

- Cells contain a small quantity of ATP and must therefore continually resynthesize it at its rate of use. Only under extreme exercise conditions do ATP levels in skeletal muscle decrease.
- A limited ATP supply provides a biologically useful mechanism to regulate energy metabolism. By maintaining only a small amount of ATP, its relative concentration changes rapidly in response to only a minimal ATP decrease.
- Any increase in energy requirement immediately disrupts the balance between ATP and ADP and Pi. As one might expect, increases in energy transfer depend on exercise intensity.
- The body stores only 80 to 100 g (about 3.0 oz) of ATP at any time under normal resting conditions. This quantity makes available each second approximately 2.4 mmol of ATP per

kg wet muscle weight, or about  $1.44 \times 10^{10}$  molecules of ATP. This represents enough intramuscular stored energy to power several seconds of explosive, all-out exercise.

- Thus, ATP alone does not represent a significant energy reserve. A sedentary person resynthesizes an amount of ATP each day equal to about 75% of body mass.
- For an endurance athlete who generates 20 times the resting energy expenditure throughout a 2.5-hour marathon race, this amounts to 80 kg of ATP resynthesis during the run! To appreciate the tremendous quantity of ATP production over the adult portion of a lifespan.