## **DME( Dropping mercury electrode)**

The **dropping mercury electrode** (**DME**) is a working electrode made of mercury and used in polarography. It is a polarisable micro-electrodes in which the drop of mercury trickles down from a resistance-glass capillary (0.05-0.08 mm in diameter and 5-9cm long) . The dropping mercury electrode has the following advantages

- (a) Its surface is reproducible, smooth, and continuously renewed; the current potential curve is reproducible and eliminates passivity or poisoning effects.
- (b) The diffusion current assumes a steady value immediately after each change of applied potential, and is reproducible.
- (c) The large hydrogen over potential on mercury allows possible the deposition of substances difficult to reduce, e.g. the alkali metal ions, aluminium ion and manganese(II) ion.
- (d) The surface area can be calculated from the weight of the drops.
  - The dropping mercury electrode is generally applied over the range +0.4 to about 2.0 volts with reference to the S.C.E. Above +0.4 volt mercury dissolves and gives an anodic wave; it begins to oxidise to mercury(I) ion.
  - At potentials more negative than about 1.8 volts with reference to the S.C.E., visible hydrogen evolution occurs in acid solutions and the usual supporting electrolytes commence to discharge.
  - The range may be extended to about 2.6 volts by using supporting electrolytes having
    higher reduction potentials than the alkali metals; tetra-alkyl ammonium hydroxides or their
    salts are satisfactory for this purpose.



