CLASSIFICATION OF REFRIGERANTS

1. Primary refrigerants

2. Secondary refrigerants

Primary refrigerants are those working mediums or heat carries which directly take part in the refrigeration system and cool the substance by the absorption of latent heat e.g., Ammonia, Carbondioride, Sulphur dioxide, Methyl chloride, Methylene chloride, Ethyl chloride and Freon group etc.

Secondary refrigerants are those circulating substances which are first cooled with the help of The primary refrigerants and are then employed for cooling purposes, e.g., ice, carbondioxide etc. These refrigerants cool substances by absorption of their sensible heat.

PRIMARY REFRIGERANTS

- 1. Halocarbon compounds
- 2. Azeotropes
- 3. Hydrocarbons
- 4. Inorganic compounds
- 5. Unsaturated organic compounds

Halocarbon Compounds refrigerants which contain one or more of three halogens, chlorine and bromine .

T hey are sold in the market under the names as Freon, Genetron, Isotron and Areton. Since the refrigerants belonging to this group have outstanding merits over the group's refrigerants, therefore they find wide field of application in domestic, commercial and industrial purposes.

Refrigerant number	Chemical name	Chemical formula
R-11	Trichloromonofluoromethane	CCI ₃ F
R-12	Dichlorodifluoromethane	CCl ₂ F ₂
R-13	Monochlorotrifluoromethane	CCIF ₃
R-14	Carbontetrafluoride	CF ₄
R-21	Dichloromonofluoromethane	CHCl ₂ F
R-22	Monochlorodifluoromethane	CHCIF ₂
R-30	Methylene chloride	CH ₂ Cl ₂
R-40	Methyl chloride	CH ₃ Cl
R-100	Ethyl chloride	C ₂ H ₅ Cl
	Trichlorotrifluorothane	CCl ₂ FCClF ₂
R-113	Dichlorotetrafluoroethane	CCIF ₂ CCIF ₂
R-114	Monochloropentafluoroethane Difluoroethane	$CCIF_2CF_3$
R-115		$C_2F_4Cl_2$
R-152	Dilluoioenano	

Azeotropes

The refrigerants are mixtures of different substances. These substances cannot be separated into components by distillation. They possess fired hermodynamic properties and do not undergo any separation with changes in temperature and pressure. An azeotrope behaves like a simple substance.

	in mining refrigerants	Chemical formula
Refrigerant number	Azeotropic mixing refrigerants	COLE CHE
	73.8% R-12 and 26.2% R-152	CCl_2F_2/CH_3CHF_2
R-500	48.8% R-22 and 51.2% R-115	CHCIF ₂ /CCIF ₂ CF ₃
R-502	40.1% R-23 and 59.9% R-13	CHF ₃ /CClF ₃
R-503		CH ₂ F ₂ /CClF ₂ CF ₃
R-504	48.2% R-32 and 51.8% R-115	

Hydrocarbons

Most of the refrigerants of this group are organic compounds. Several hydrocarbons are used successfully in commercial and industrial installations. Most of them possess satisfactory thermodynamic properties but are highly inflammable

Refrigerant number	Chemical name	Chemical formula
R-50	and the second second second second second second	
R-170	Methane	CH ₄
R-290	Ethane	C ₂ H ₆
	Propane	C_2H_8
R-600	Butane	C_4H_{10}
R-601	Isobutane	[CH(CH ₃) ₃]

Inorganic Compounds

Before the introduction of hydrocarbon group these refrigerants were most commonly used for all purposes.

Refrigerant number	Chemical name	Chemical formula
R-717	Ammonia	NH ₃
R-718	Water Air	H ₂ O
R-729		Mixturer of
		O_2 , N_2 , CO_2 etc
R -744	Carbon dioxide	CO ₂
R-764	Sulphur dióxide	SO ₂

Unsaturated Organic Compounds

The refrigerants belonging to this group possess ethylene or propylene as their constituents.

Refrigerant name	Chemical name
R-1120	Trichloroethylene
R-1130	Dichloroethylene
R -1150	Ethylene
R-1270	Propylene