

# Definition of structures

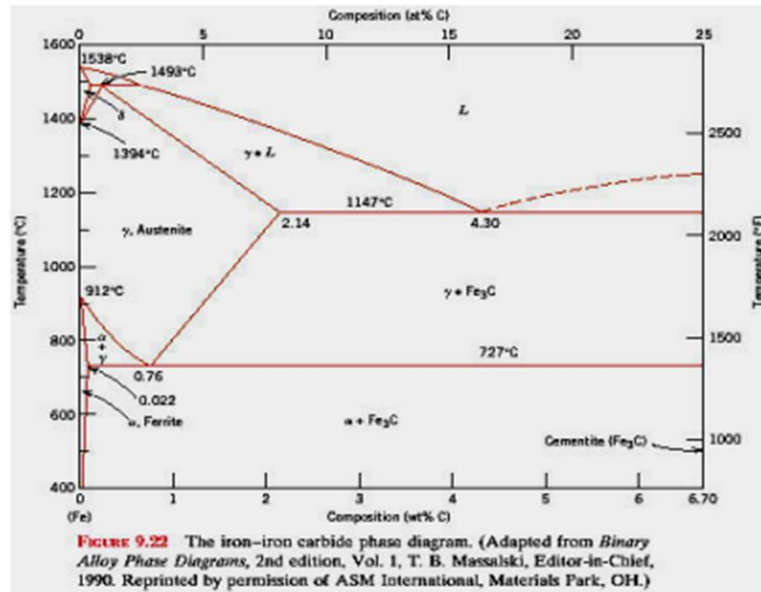
- **Ledeburite** is the eutectic mixture of austenite and cementite.
- It contains 4.3 percent C and is formed at 1130°C.

**Martensite** - a super-saturated solid solution of carbon in ferrite.

It is formed when steel is cooled so rapidly that the change from austenite to pearlite is suppressed.

The interstitial carbon atoms distort the BCC ferrite into a BC-tetragonal structure (BCT).; responsible for the hardness of quenched steel

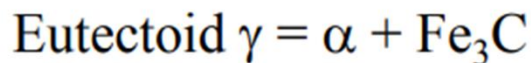
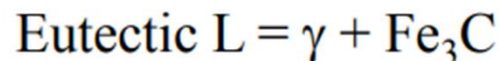
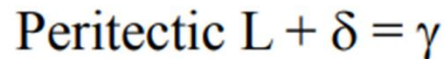
# Various Features of Fe-C diagram



## Phases present

- Liquid (l)
- $\delta$ : BCC structure, Paramagnetic
- $\alpha$  ferrite: BCC structure, Ferromagnetic, Fairly ductile
- $\gamma$  austenite: FCC structure, Non-magnetic, Ductile
- $\text{Fe}_3\text{C}$  cementite: Orthorhombic, Hard, brittle

## Reactions



Max. solubility of C in ferrite = 0.022%

Max. solubility of C in austenite = 2.11%

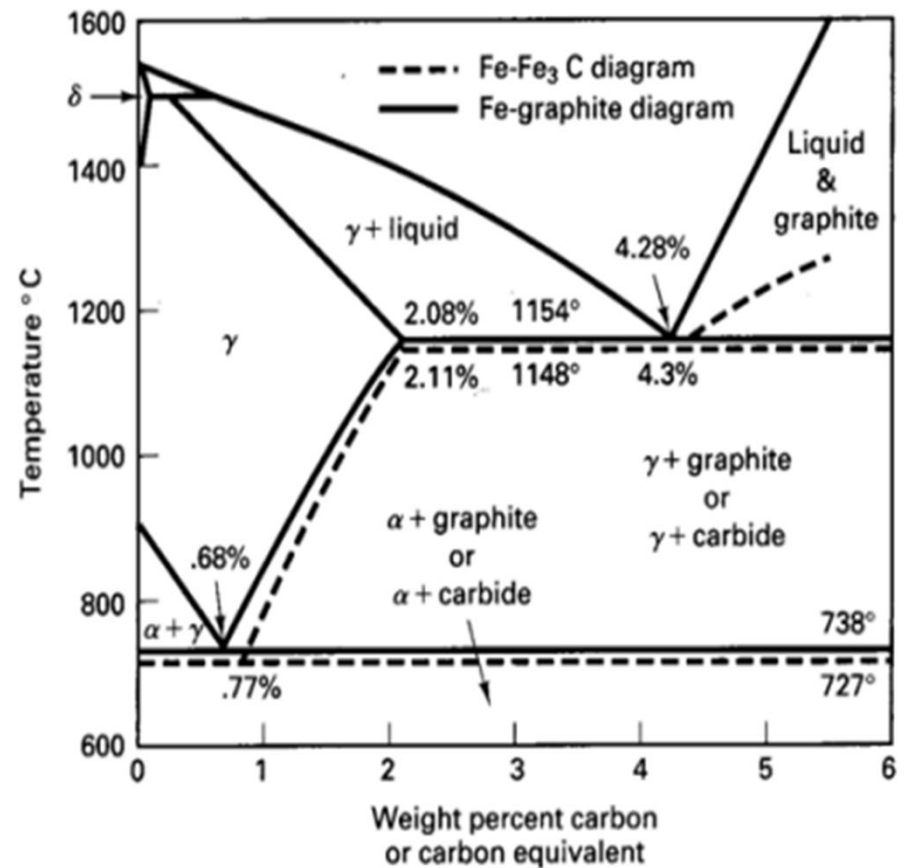
# Three Phase Reactions

A horizontal line always indicates an invariant reaction in binary phase diagrams

- Peritectic reaction at  $1495^{\circ}\text{C}$  and  $0.18\% \text{C}$ ,
  - $\delta\text{-ferrite} + L \leftrightarrow \gamma\text{-iron (austenite)}$
  - (almost no engineering importance).
- Eutectic reaction at  $1147^{\circ}\text{C}$  and  $4.3\% \text{C}$ ,
  - $L \leftrightarrow \gamma\text{-iron} + \text{Fe}_3\text{C (cementite)}$  [ledeburite]
  - alloys called **cast irons**
- Eutectoid reaction at  $727^{\circ}\text{C}$  and  $0.77\% \text{C}$ ,
  - $\gamma\text{-iron} \leftrightarrow \alpha\text{-ferrite} + \text{Fe}_3\text{C (cementite)}$  [pearlite]
  - They are **steels**

# Cast Irons

- Iron-Carbon alloys of 2.11%C or more are cast irons.
- Typical composition: 2.0-4.0%C, 0.5-3.0% Si, less than 1.0% Mn and less than 0.2% S.
- Si-substitutes partially for C and promotes formation of graphite as the carbon rich component instead  $\text{Fe}_3\text{C}$ .



# The Austenite to ferrite / cementite transformation in relation to Fe-C diagram

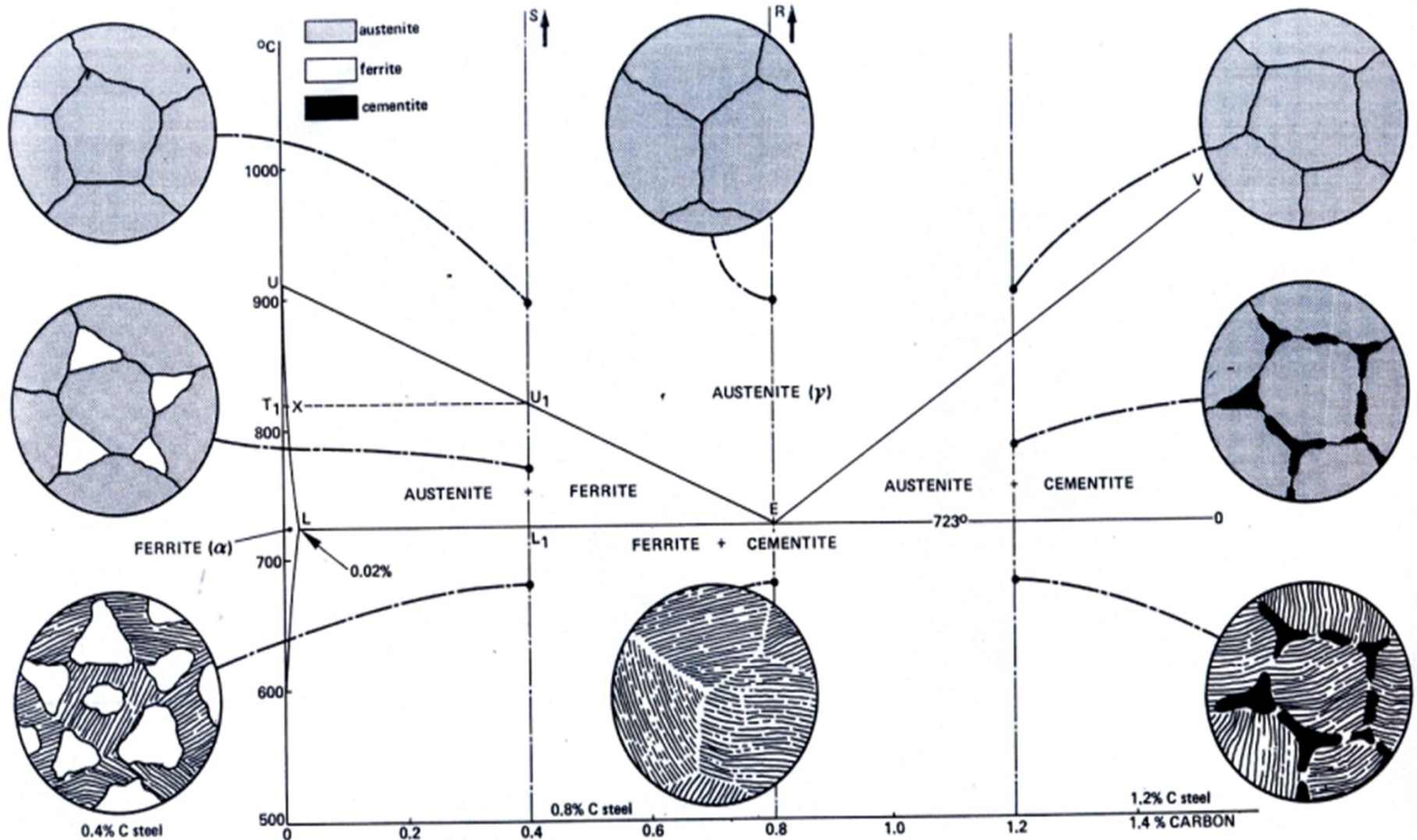


Fig. 9.3—The austenite  $\rightarrow$  ferrite/cementite transformation in relation to the iron-carbon diagram.

## Principal phases of steel and their Characteristics

<b>Phase</b>	<b>Crystal structure</b>	<b>Characteristics</b>
<b>Ferrite</b>	<b>BCC</b>	<b>Soft, ductile, magnetic</b>
<b>Austenite</b>	<b>FCC</b>	<b>Soft, moderate strength, non-magnetic</b>
<b>Cementite</b>	<b>Compound of Iron &amp; Carbon Fe<sub>3</sub>C</b>	<b>Hard &amp; brittle</b>