

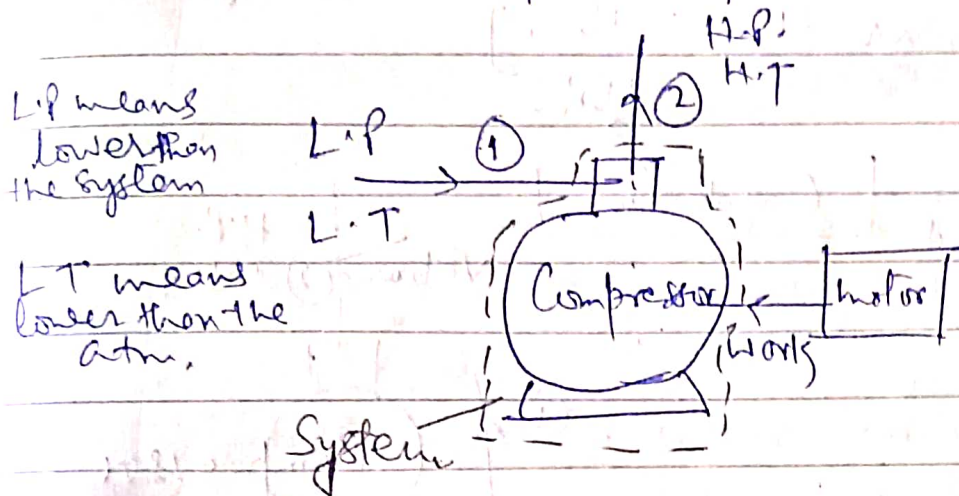
$$\therefore \frac{T_2}{T_1} = \frac{T_3}{T_4} = \frac{T_2 - T_3}{T_1 - T_4} = r_p^{\frac{\gamma-1}{\gamma}}$$

$$\therefore \left(\text{C.O.P} \right) = \frac{1}{r_p^{\left(\frac{\gamma-1}{\gamma} \right)} - 1}$$

Bell
Coleman
cycle

Vapour Compression refrigeration system

In this system vapour refrigerant are compressed to such a level that heat is rejected to the atmosphere, hence it is called as such Vapour Compression refrigeration system.



Process 1-2 Isentropic Compression in a Compressor.

Energy entering = Energy outgoing

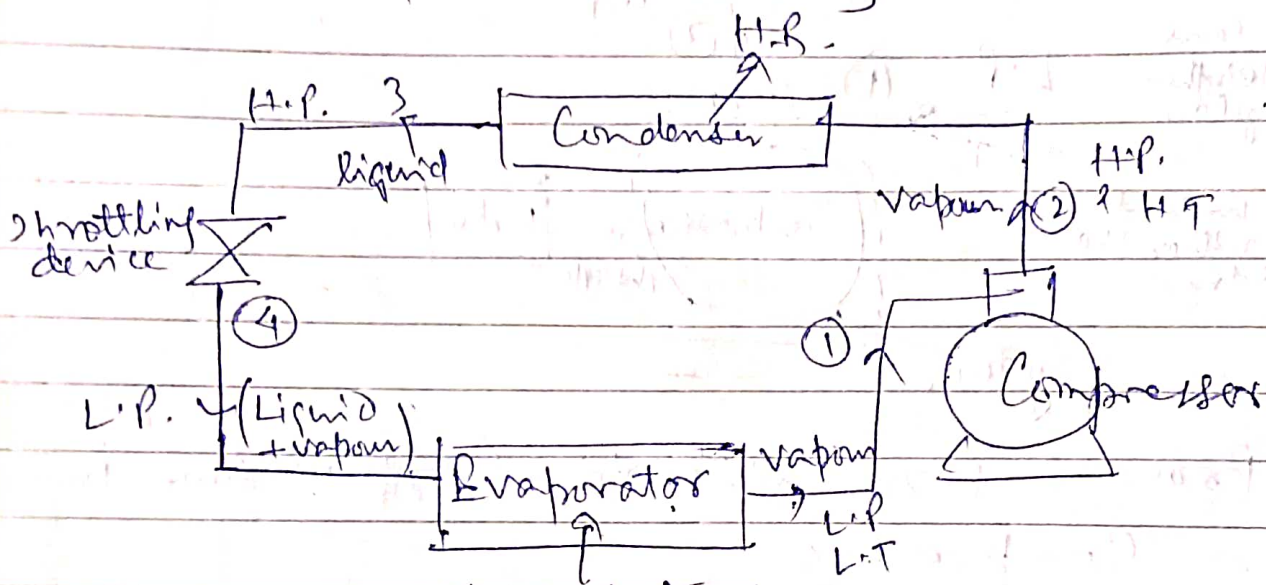
$$h_1 + W_c = h_2$$

$$W_c = h_2 - h_1$$

Compressor + motor sealed in glove and surrounded by electrical windings. Then it is known as

- (i) ~~sealed unit~~ Sealed unit or Hermetically sealed unit
- (ii) Semi sealed unit (Dome is surrounded by cover plate & tightened by nut bolt)
- (iii) open unit.

[Refrigerant vapour temp is lower than atm. ~~pressure~~ temp. If we compress it i.e., increase its pr. then corresponding temp increases, therefore we compresses such a level that vapour temp is higher than atmosphere. Hence heat flow from vapour to atm.]



Process 2-3: Heat abstracted
 Constant pressure heat rejection
 in the Condenser.

$$h_2 = h_{h.} + h_3$$

$$h_{h.} = h_2 - h_3$$

In refrigerator, this cooling is obtained with the help of air then it is called as Air Cooled Condenser.

Another type is water Cooled Condenser used in Ice plant, Building refrigeration

Process 3-4: Throttling process.

$$h_3 = h_4$$

Due to friction, some liquid will be converted into vapour. Therefore after throttling we get mix. of liquid and vapour.

Throttling device could be a

1. Capillary-tube

($\frac{1}{30}$ th to $\frac{1}{40}$ th of an inch)

It is hold good for refrigerator and air conditioning.

2. Thermostatic exp. valve

3. Automatic exp. valve

4. Float valve $\begin{matrix} \rightarrow \text{H.P. side} \\ \rightarrow \text{L.P. side} \end{matrix}$

Process 4-1:

only liquid is evaporated into vapour.

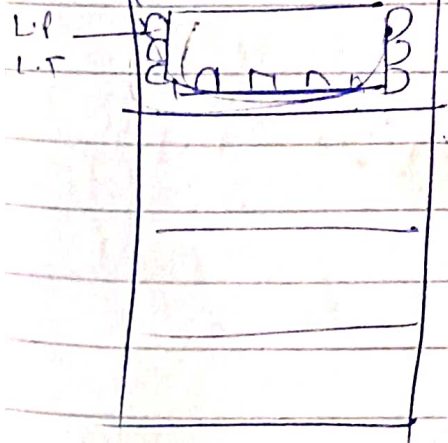
Technically called evaporator.

In refrigerator (In air conditioning) freezer. we collect cooling coil

Constant pressure heat

abstracted in the evaporator

(Net refrigerating effect, N)

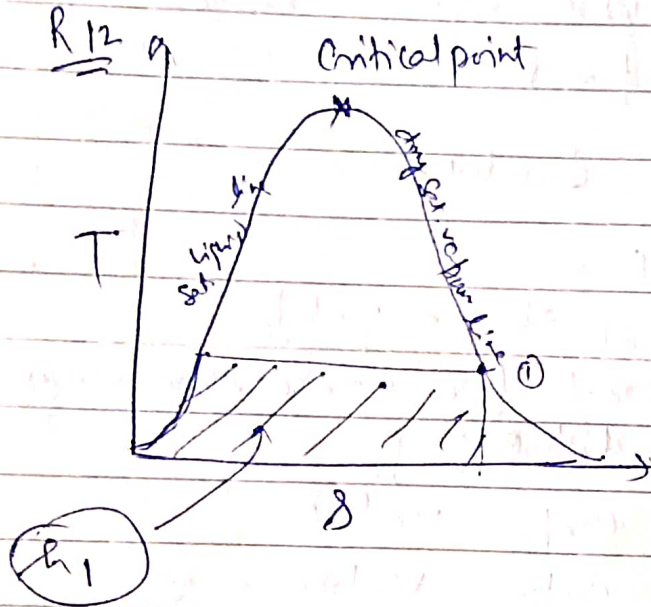


$$h_4 + N = h_1$$

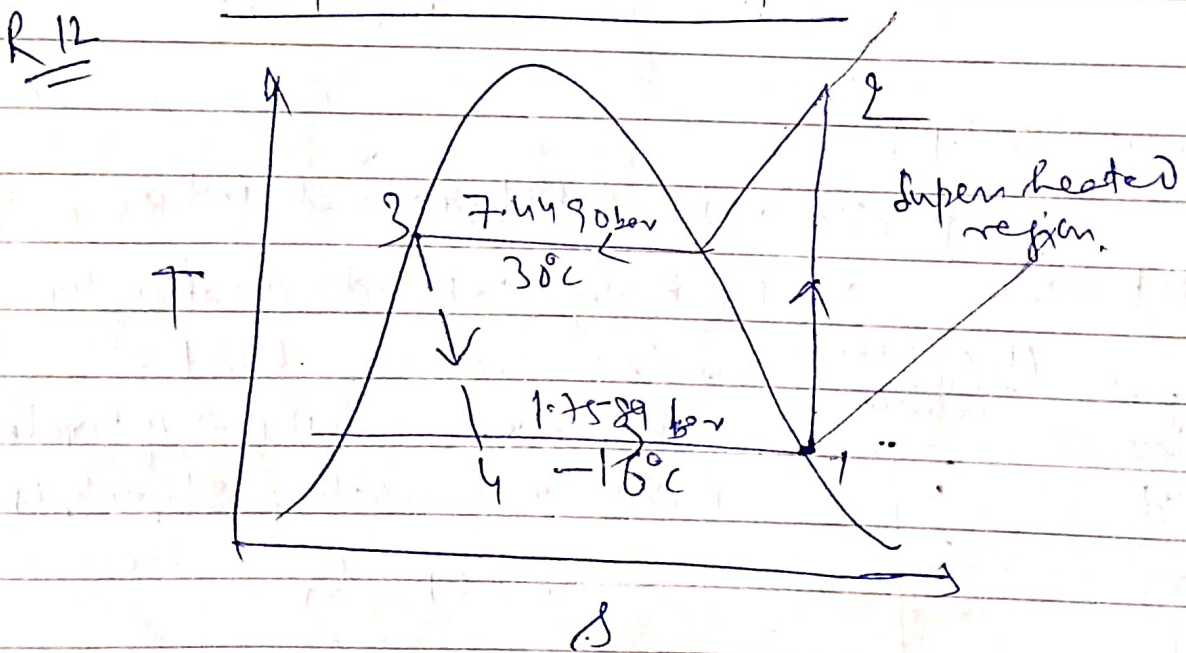
$$N = h_1 - h_4$$

Evaporator

- (1) Freezer (refrigerator)
- (2) Cooling coil (A/C)
- (3) Chiller (Ice plant)

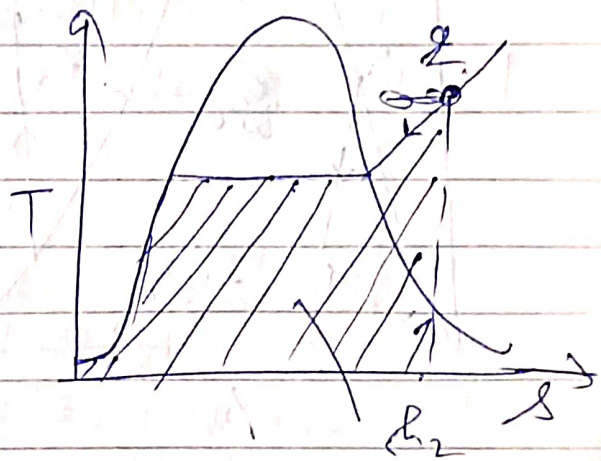


Simple Saturation Cycle



for finding enthalpy of any point,

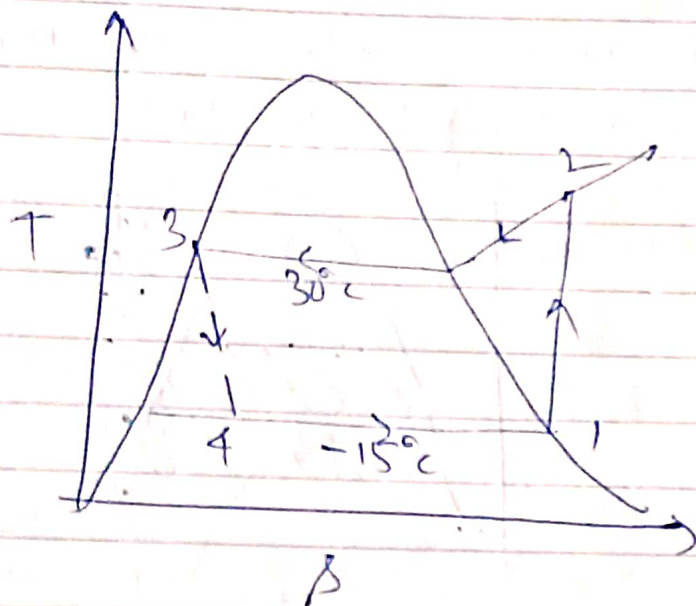
- (i) from the pt. under consideration draw a vertical line.
- (ii) from the left hand side of the point move along the process line till we reach the liquid line.
- (iii) Move along the liquid line till datum.
- (iv) This entire area represents the enthalpy at the point under consideration.



Q.1) Def. a refrigerator working on a ^{STD} vapor compression cycle, ϕ of evaporator & condenser temp are -15°C and 30°C then calculate h_1 , work reqd, N , C.O.P, Heat rejected in the Condenser, using R12.

Also calculate power and mass flow rate if capacity is 5 TR.

Soln
R12



$$h_1 = (h_g)_{-15^\circ\text{C}} = 344.92 \frac{\text{kJ}}{\text{kg}}$$

$$s_1 = s_2 = 1.56317 \frac{\text{kJ}}{\text{kgK}}$$

In superheated vapour,

At 30°C (or 7.4490 bar)

35°C	\longrightarrow	1.555
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?	\longrightarrow	1.56317
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40°C	\longrightarrow	1.5673
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$$\rightarrow = 35 + \frac{5}{(1.5673 - 1.555)} \times (1.56317 - 1.555)$$

$$= 38.25^\circ\text{C}$$

Temp ^o	Enthalpy
35 ^o c	→ 367.274
38.25 ^o c	→ ?
40 ^o c	→ 370.944

$$h_2 = 367.274 + \frac{(370.944 - 367.274)}{5} \times 3.25$$

$$h_2 = 368.659 \text{ kJ/kg}$$

$$h_3 - h_4 = (h_f)_{50^{\circ}\text{C}} = 228.54 \text{ kJ/kg}$$

$$\begin{aligned} \text{Work reqd} &= h_2 - h_1 \\ &= 368.659 - 344.92 \\ &= 24.739 \text{ kJ/kg} \end{aligned}$$

Net refrigerating effect

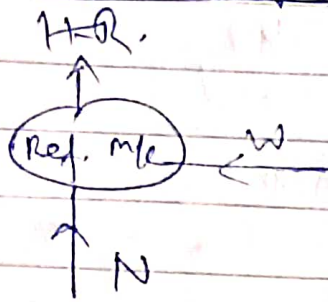
$$\begin{aligned} (N) &= h_1 - h_4 \\ &= 344.92 - 228.54 \\ &= 116.38 \text{ kJ/kg} \end{aligned}$$

$$\text{C.O.P} = \frac{N}{W} = \frac{116.38}{24.739} = 4.70$$

Heat rejected in the Condenser

$$\begin{aligned} &= h_2 - h_3 \\ &= 368.659 - 228.54 = 141.119 \text{ kJ/kg} \end{aligned}$$

For check up



$$N + W = H.R.$$

$$116.38 + 24.739 = 141.118$$

Given Capacity = 5 TR.

$$m(h_1 - h_4) = 5 \times 3.5$$

$$m = \frac{5 \times 3.5}{116.38} = 0.1504 \text{ kg/s}$$

$$m = 0.1504 \text{ kg/s}$$

$$\begin{aligned} \text{Power reqd.} &= m(h_2 - h_1) \\ &= 0.1504 \times 24.739 \\ &= 3.7207 \text{ kW} \end{aligned}$$

②

It is proposed to replace R12 by ozone friendly R134a in a refrigeration plant of 10 TR Capacity with evaporator and Condenser temp of 0°C and 40°C respectively. Considering standard saturation cycle (evaporator exit) and Condense exit as saturated stage, Compare the mass flow rate, Compressor work (kW), Condenser heat rejection (kW) & COP for the two refrigerants. The

Saturation properties and vapour specific heat are as follows

$t^{\circ}\text{C}$	h_f ($\frac{\text{kJ}}{\text{kg}}$)	h_g ($\frac{\text{kJ}}{\text{kg}}$)	s_f ($\frac{\text{kJ}}{\text{kgK}}$)	s_g ($\frac{\text{kJ}}{\text{kgK}}$)	C_{p0} ($\frac{\text{kJ}}{\text{kg}^{\circ}\text{C}}$)
134	0	200.81	398.78	1.0025	—
40	255.73	419.63	1.1884	1.7128	1.068
R12	0	361	187.5	0.0142	—
40	74.6	283.2	0.2718	0.6847	0.776

Soln.

R134a

$$h_1 = h_{g1} = 398.78 \frac{\text{kJ}}{\text{kg}}$$

$$s_1 = s_2$$

$$(s_g)_1 = (s_{\text{sup}})_2$$

$$1.7261 = \left[s_g + C_p \ln \frac{T_2}{T_8} \right]_2$$

$$= 1.7128 + 1.068 \ln \frac{T_2}{313}$$

$$\therefore T_2 = 316.92 \text{ K } (43.92^{\circ}\text{C})$$

$$h_2 = h_g + C_p (T_2 - T_8)$$

$$= 419.63 + 1.068 (43.92 - 40)$$

$$\boxed{h_2 = 423.82} \frac{\text{kJ}}{\text{kg}}$$

